## SADAKATHULLAH APPA COLLEGE

(AUTONOMOUS)

(Reaccredited by NAAC with an 'A' Grade with a CGPA of 3.40 out of 4.00 in the III cycle An ISO 9001:2015 Certified Institution)

RAHMATH NAGAR, TIRUNELVELI- 11.

Tamilnadu



## **CBCS SYLLABUS**

For

**M.Sc. CHEMISTRY** 

(As per the Resolutions of the Academic Council Meeting held on 03.03.2018 & 17.10.2018)

## CONTENTS

| Sl.<br>No. | Subject Title                    | Subject Code | Page<br>No. |
|------------|----------------------------------|--------------|-------------|
| 1          | Course Structure                 | -            | 1           |
| 2          | INORGANIC CHEMISTRY – I          | 18PCCH11     | 6           |
| 3          | ORGANIC CHEMISTRY – I            | 18PCCH12     | 7           |
| 4          | PHYSICAL CHEMISTRY – I           | 18PCCH13     | 8           |
| 5          | CHROMATOGRAPHY                   | 18PECH1A     | 9           |
| 6          | BIOCHEMISTRY                     | 18PECH1B     | 10          |
| 7          | INORGANIC CHEMISTRY PRACTICAL -I | 18PCCH1P1    | 11          |
| 8          | PHYSICAL CHEMISTRY PRACTICAL - I | 18PCCH1P2    | 12          |
| 9          | INORGANIC CHEMISTRY – II         | 18PCCH21     | 13          |
| 10         | ORGANIC CHEMISTRY – II           | 18PCCH22     | 14          |
| 11         | PHYSICAL CHEMISTRY – II          | 18PCCH23     | 15          |
| 12         | APPLIED CHEMISTRY                | 18PECH2A     | 16          |
| 13         | CHEMINFORMATICS                  | 18PECH2B     | 17          |
| 14         | ORGANIC CHEMISTRY PRACTICAL – I  | 18PCCH2P1    | 18          |
| 15         | CHROMATOGRAPHY PRACTICAL         | 18PCCH2P2    | 19          |
| 16         | INORGANIC CHEMISTRY – III        | 18PCCH31     | 20          |
| 17         | ORGANIC CHEMISTRY – III          | 18PCCH32     | 21          |
| 18         | PHYSICAL CHEMISTRY – III         | 18PCCH33     | 22          |
| 19         | INSTRUMENTAL METHODS OF ANALYSIS | 18PECH3A     | 23          |
| 20         | ENZYME CHEMISTRY                 | 18PECH3B     | 24          |
| 21         | ORGANIC CHEMISTRYPRACTICAL - II  | 18PCCH3P1    | 25          |
| 22         | PHYSICAL CHEMISTRY PRACTICAL-II  | 18PCCH3P2    | 26          |
| 23         | SPECTROSCOPY                     | 18PCCH41     | 27          |
| 24         | ADVANCED TOPICS IN CHEMISTRY     | 18PCCH42     | 28          |
| 25         | PROJECT                          | 18PCCH43     | 29          |
| 26         | MEDICINAL CHEMISTRY              | 18PECH4A     | 30          |
| 27         | RATIONAL DRUG DESIGN             | 18PECH4B     | 31          |
| 28         | INORGANIC CHEMISTRY PRACTICAL II | 18PCCH4P1    | 32          |
| 29         | GREEN CHEMISTRY PRACTICAL        | 18PCCH4P2    | 33          |
| 30         | INDUSTRIAL CHEMISTRY             | 18PICH21     | 34          |
| 31         | INTRODUCTION TO CHEMINFORMATICS  | 18PICH31     | 35          |

| (Applicable for students admitted in June 2018 and onwards) |                              |                                     |             |       |     |       |    |          |  |
|---|------------------------------|-------------------------------------|-------------|-------|-----|-------|----|----------|--|
|   | Department of Chemistry (PG) |                                     |             |       |     |       |    |          |  |
|   | C                            | BCS Syllabus – M.Sc., Chem          | istry (2018 | -19 o | nw  | ards  | )  |          |  |
| Sem P   |                              | Title of the Paper                  | Sub. Code   | H/W   | С   | Marks |    | KS –     |  |
|   | D.G.G.1                      |                                     | 100001111   |       |     | I     | E  | <b>T</b> |  |
|   | DSC1                         | INORGANIC CHEMISTRY – I             | 18PCCH11    | 6     | 4   | 25    | 75 | 100      |  |
|   | DSC2                         | ORGANIC CHEMISTRY – I               | 18PCCH12    | 6     | 4   | 25    | 75 | 100      |  |
|   | DSC3                         | PHYSICAL CHEMISTRY – I              | 18PCCH13    | 6     | 4   | 25    | 75 | 100      |  |
|   | DSE1A                        | CHROMATOGRAPHY                      | 18PECH1A    | 4     | 4   | 25    | 75 | 100      |  |
| Ι   | DSE1B                        | BIOCHEMISTRY                        | 18PECH1B    | •     |     | 20    | 10 | 100      |  |
|   | P-I                          | INORGANIC CHEMISTRY<br>PRACTICAL –I | 18PCCH1P1   | 4     | 2   | 40    | 60 | 100/2    |  |
|   | P-II                         | PHYSICAL CHEMISTRY<br>PRACTICAL – I | 18PCCH1P2   | 4     | 2   | 40    | 60 | 100/2    |  |
|   | DSC4                         | INORGANIC CHEMISTRY – II            | 18PCCH21    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSC5                         | ORGANIC CHEMISTRY – II              | 18PCCH22    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSC6                         | PHYSICAL CHEMISTRY – II             | 18PCCH23    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSE2A                        | APPLIED CHEMISTRY                   | 18PECH2A    | 4     | 4   | 05    |    | 100      |  |
|   | DSE2B                        | CHEMINFORMATICS                     | 18PECH2B    | 4     | 4   | 25    | 75 | 100      |  |
| 11  | P-III                        | ORGANIC CHEMISTRY<br>PRACTICAL – I  | 18PCCH2P1   | 4     | 2   | 40    | 60 | 100/2    |  |
|   | P-IV                         | CHROMATOGRAPHY<br>PRACTICAL         | 18PCCH2P2   | 4     | 2   | 40    | 60 | 100/2    |  |
|   | IDC-1                        | INDUSTRIAL CHEMISTRY                | 18PICH21    | 3     | 3   | 25    | 75 | 100      |  |
|   | DSC7                         | INORGANIC CHEMISTRY – III           | 18PCCH31    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSC8                         | ORGANIC CHEMISTRY – III             | 18PCCH32    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSC9                         | PHYSICAL CHEMISTRY – III            | 18PCCH33    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSE3A                        | INSTRUMENTAL METHODS<br>OF ANALYSIS | 18PECH3A    | 4     | 4   | 4 25  | 75 | 100      |  |
| ш   | DSE3B                        | ENZYME CHEMISTRY                    | 18PECH3B    | ,     | + + |       |    |          |  |
|   | P-V                          | ORGANIC<br>CHEMISTRYPRACTICAL – II  | 18PCCH3P1   | 4     | 2   | 40    | 60 | 100/2    |  |
|   | P-VI                         | PHYSICAL CHEMISTRY<br>PRACTICAL-II  | 18PCCH3P2   | 4     | 2   | 40    | 60 | 100/2    |  |
|   | IDC-2                        | INTRODUCTION TO<br>CHEMINFORMATICS  | 18PICH31    | 3     | 3   | 25    | 75 | 100      |  |
|   | DSC10                        | SPECTROSCOPY                        | 18PCCH41    | 5     | 4   | 25    | 75 | 100      |  |
|   | DSC11                        | ADVANCED TOPICS IN<br>CHEMISTRY     | 18PCCH42    | 5     | 4   | 25    | 75 | 100      |  |
| IV  | DSC12                        | PROJECT                             | 18PCCH43    | 8     | 8   |       |    | 100      |  |
|   | DSE4A                        | MEDICINAL CHEMISTRY                 | 18PECH4A    |       | -   |       |    |          |  |
|   | DSE4B                        | RATIONAL DRUG DESIGN                | 18PECH4B    | 4     | 4   | 25    | 75 | 100      |  |
|   | P-VII                        | INORGANIC CHEMISTRY                 | 18PCCH4P1   | 4     | 2   | 40    | 60 | 100/2    |  |

| ( <i>I</i>               | (Applicable for students admitted in June 2018 and onwards)<br>Department of Chemistry (PG)<br>CBCS Syllabus – M.Sc., Chemistry (2018-19 onwards) |                 |           |            |    |       |    |       |       |
|--------------------------|---|-----------------|-----------|------------|----|-------|----|-------|-------|
| Sem P Title of the Paper |   |                 | Sub. Code | н/w        | С  | Marks |    |       |       |
|                          | -   |                 |           | ,          | C  | Ι     | E  | T     |       |
|                          |   | PRACTICAL II    |           |            |    |       |    |       |       |
|                          |   | GREEN CHEMISTRY | 100001400 | 1000011400 | 4  | 0     | 10 | 60    | 100/0 |
|                          | P-VIII  | PRACTICAL       | 10PCCH4P2 | 4          | 4  | 40    | 00 | 100/2 |       |
|                          |   | Total           |           | 120        | 90 |       |    | 2200  |       |

| PO -<br>No. | Upon completion of M. Sc., Degree program, the post graduates will be able to:   |
|-------------|--|
| PO-1        | Understand the fundamentals and application of chemical and<br>scientific theories in Inorganic, Organic, Physical and Biological<br>chemistry.  |
| PO-2        | Build in the skills of problem solving, critical thinking and analytical reasoning from handling the scientific problems.  |
| PO-3        | Create an appropriate model, formulate a hypothesis, choose an<br>appropriate set of tools and techniques and design an experiment<br>that tests the hypothesis and analyze the results from the<br>experiment based on sound scientific conclusions from the results<br>obtained. |
| PO-4        | Understand the role of science in the society and imbibe ethical<br>values of safe handling of chemicals, environmental issues and key<br>issues that the society is facing in energy, health and medicine.  |
| PO-5        | Analyze the practical problems and think creatively to generate innovative solutions using appropriate technologies.   |
| PO-6        | Develop the interpersonal skills to function cooperatively in a team setting.  |
| PO-7        | Apply effective scientific communication in oral and writing, for scientific concepts, proposals and monographs, <i>etc.</i> ,   |

| PSO – | Upon completion of M. Sc., Chemistry Degree                   | PO     |
|-------|---|--------|
| No.   | program, the post graduates will be able to:                  | Mapped |
| PSO-1 | Develop broad set of chemical concepts concerning the         | PO-1   |
|       | fundamentals in the basic areas of the discipline             |        |
|       | (Organic, Inorganic, Physical and Analytical chemistry).      |        |
| PSO-2 | Understand the chemical reactions, structures and properties. | PO-2   |
| PSO-3 | Apply the knowledge of fundamentals of chemistry to           | PO-3   |
|       | find the solution for chemical problems.                      |        |
| PSO-4 | Formulate hypotheses and assess the chemical concepts         | PO-4   |
|       | using skills learnt in laboratories to perform                |        |
|       | experiments, collect data, compile and interpret results      |        |
|       | and draw reasonable and logical conclusions.                  |        |
| PSO-5 | Apply the skills in handling, synthesizing, purifying and     | PO-5   |
|       | characterizing new and existing chemicals which include       |        |
|       | knowing the proper procedures and regulations for the         |        |
|       | safe handling, use and disposal of chemicals.                 |        |
| PSO-6 | Develop appropriate techniques, resources and modern          | PO-7   |
|       | software tools including prediction and modelling to          |        |
|       | chemical science activities.                                  |        |
| PSO-7 | Understand the benefits and impacts in the society and        | PO-4   |
|       | apply it as a basis for ethical behaviour in solving the      |        |
|       | problems faced by the chemists. Safe handle the               |        |
|       | chemicals, solve environmental issues and key issues of       |        |
|       | the society regarding energy, health and medicine.            |        |
| PSO-8 | Develop the ability to function as an individual and          | PO-6   |
|       | member of an interdisciplinary problem solving team.          |        |

| Course               | Details               |    |   |            |   |  |
|----------------------|-----------------------|----|---|------------|---|--|
| Code                 | 18PCCH11              |    |   |            |   |  |
| Title                | INORGANIC CHEMISTRY-I |    |   |            |   |  |
| Degree               | M. Sc.,               |    |   |            |   |  |
| Branch(s)            | Chemistry             |    |   |            |   |  |
| Year/Semester        | Ι                     |    | Ι |            |   |  |
| Туре                 | Core                  |    |   |            |   |  |
| Credits              | 4                     |    |   |            |   |  |
| No. of Contact Hours | Total Hours           | 90 |   | Hours/Week | 6 |  |

| CO-No. | Upon completion of this course, students will be able to:                          | PSO<br>addressed | Cognitive Level |
|--------|--|------------------|-----------------|
| CO-1   | Understand and categorise the types of crystal structure.                          | PSO-1            | Understanding   |
| CO-2   | Construct the molecular orbital diagram for Homo, Hetero and polyatomic molecules. | PSO-3            | Applying        |
| CO-3   | Analyze the stereochemical<br>properties of coordination<br>complexes.             | PSO-4            | Analyzing       |
| CO-4   | Understanding the crystal field theory and its splitting behaviour.                | PSO-6            | Understanding   |
| CO-5   | Interpret the stability of coordination complexes.                                 | PSO-5            | Evaluating      |

| Course               | Details               |    |            |   |  |  |
|----------------------|-----------------------|----|------------|---|--|--|
| Code                 | 18PCCH12              |    |            |   |  |  |
| Title                | ORGANIC CHEMISTRY – I |    |            |   |  |  |
| Degree               | M. Sc.,               |    |            |   |  |  |
| Branch(s)            | Chemistry             |    |            |   |  |  |
| Year/Semester        | Ι                     | Ι  |            |   |  |  |
| Туре                 | Core                  |    |            |   |  |  |
| Credits              | 4                     |    |            |   |  |  |
| No. of Contact Hours | Total Hours           | 90 | Hours/Week | 6 |  |  |

| CO-No. | Upon completion of the course, students will able to:   | PSO<br>addressed | Cognitive Level              |
|--------|---|------------------|------------------------------|
| CO – 1 | Understand the reactions and mechanisms.  | PSO-1            | Understanding                |
| CO – 2 | Apply suitable oxidising and reducing<br>reagents for different organic<br>reactions.             | PSO-1,2          | Applying                     |
| CO – 3 | Apply suitable organic reagents for<br>designing organic transformations in<br>organic compounds. | PSO-7            | Applying                     |
| CO – 4 | Explain the reaction mechanism for organic reactions.   | PSO-2            | Understanding,<br>Evaluating |
| CO – 5 | Interpret the ring structure and stability.   | PSO-3            | Evaluating                   |

| Course               | Details                |    |            |   |  |  |
|----------------------|------------------------|----|------------|---|--|--|
| Code                 | 18PCCH13               |    |            |   |  |  |
| Title                | PHYSICAL CHEMISTRY – I |    |            |   |  |  |
| Degree               | M. Sc.,                |    |            |   |  |  |
| Branch(s)            | Chemistry              |    |            |   |  |  |
| Year/Semester        | Ι                      | Ι  |            |   |  |  |
| Туре                 | Core                   | ·  |            |   |  |  |
| Credits              | 4                      |    |            |   |  |  |
| No. of Contact Hours | Total Hours            | 90 | Hours/Week | 6 |  |  |

| CO-No. | Upon completion of this course, students will be able to:            | PSO<br>addressed | Cognitive Level              |
|--------|--|------------------|------------------------------|
| CO-1   | Understand and explain the concept<br>of thermodynamics properties.  | PSO-1,7          | Understanding                |
| CO-2   | Apply the concept of irreversible thermodynamics to systems.         | PSO-2            | Applying                     |
| CO-3   | Explain the activated complex theories and rate constant.            | PSO-3            | Understanding,<br>Evaluating |
| CO-4   | Interpret the properties of wave function.                           | PSO-8            | Evaluating                   |
| C0-5   | Apply Schrodinger equations to different systems and find solutions. | PSO-5            | Applying                     |

| Course               | Details      |    |            |   |
|----------------------|--------------|----|------------|---|
| Code                 | 18PECH1A     |    |            |   |
| Title                | CHROMATOGRAP | НҮ |            |   |
| Degree               | M. Sc.,      |    |            |   |
| Branch(s)            | Chemistry    |    |            |   |
| Year/Semester        | Ι            | Ι  |            |   |
| Туре                 | Elective     |    |            |   |
| Credits              | 4            |    |            |   |
| No. of Contact Hours | Total Hours  | 60 | Hours/Week | 4 |

| CO-No. | Upon completion of this course, students will be able to:                                  | PSO addressed | Cognitive Level              |
|--------|--|---------------|------------------------------|
| CO-1   | Explain the chromatographic techniques.  | PSO-1         | Understanding,<br>Evaluating |
| CO-2   | Illustrate the principle and uses of thin layer chromatography.                            | PSO-4         | Understanding                |
| CO-3   | Compare the cation and anion<br>exchange resins in ion exchange<br>chromatography.         | PSO-6         | Understanding                |
| CO-4   | Interpret the stationary and mobile<br>phase in high performance liquid<br>chromatography. | PSO-3,6       | Evaluating                   |
| CO-5   | Summarize the techniques and applications of gas chromatography.                           | PSO-8         | Understanding                |

| Course               | Details      |              |            |   |  |  |
|----------------------|--------------|--------------|------------|---|--|--|
| Code                 | 18PECH1B     |              |            |   |  |  |
| Title                | BIOCHEMISTRY | BIOCHEMISTRY |            |   |  |  |
| Degree               | M. Sc.,      |              |            |   |  |  |
| Branch(s)            | Chemistry    |              |            |   |  |  |
| Year/Semester        | Ι            | Ι            |            |   |  |  |
| Туре                 | Elective     |              |            |   |  |  |
| Credits              | 4            |              |            |   |  |  |
| No. of Contact Hours | Total Hours  | 60           | Hours/Week | 4 |  |  |

| CO-No. | Upon completion of the course, students will able to:                | PSO<br>addressed | Cognitive Level              |
|--------|--|------------------|------------------------------|
| CO- 1  | Identify the structures of carbohydrates.                            | PSO-1,2          | Applying                     |
| CO - 2 | Classify the amino acids and proteins                                | PSO-1,3          | Understanding                |
| CO - 3 | Classify different lipids and<br>understand its biological functions | PSO-4            | Understanding                |
| CO – 4 | Outline the structures and functions of nucleic acid                 | PSO-6            | Understanding                |
| CO – 5 | Explain the metabolic reactions.                                     | PSO-8            | Understanding,<br>Evaluating |

| Course               | Details                           |    |            |   |  |
|----------------------|-----------------------------------|----|------------|---|--|
| Code                 | 18PCCH1P1                         |    |            |   |  |
| Title                | INORGANIC CHEMISTRY PRACTICAL - I |    |            |   |  |
| Degree               | M. Sc.,                           |    |            |   |  |
| Branch(s)            | Chemistry                         |    |            |   |  |
| Year/Semester        | Ι                                 | Ι  |            |   |  |
| Туре                 | Core Practical                    | ·  |            |   |  |
| Credits              | 2                                 |    |            |   |  |
| No. of Contact Hours | Total Hours                       | 60 | Hours/Week | 4 |  |

| CO-No. | Upon completion of this course, students will be able to:                      | PSO addressed | Cognitive Level |
|--------|--|---------------|-----------------|
| CO-1   | Determine the less familiar cations in semi micro analysis.                    | PSO-3,4       | Evaluating      |
| CO-2   | Analyze the familiar cations in the salt mixture.                              | PSO-6,7       | Analyzing       |
| CO-3   | Design the estimation of inorganic<br>elements by complexometric<br>titration. | PSO-8         | Creating        |
| CO-4   | Outline the methods of analysis.   | PSO-4         | Understanding   |
| CO-5   | Design the methods for analyzing the salt mixtures.                            | PSO-3         | Creating        |

| Course               | Details                        |    |            |   |  |
|----------------------|--------------------------------|----|------------|---|--|
| Code                 | 18PCCH1P2                      |    |            |   |  |
| Title                | PHYSICAL CHEMISTRY PRACTICAL I |    |            |   |  |
| Degree               | M. Sc.,                        |    |            |   |  |
| Branch(s)            | Chemistry                      |    |            |   |  |
| Year/Semester        | Ι                              | Ι  |            |   |  |
| Туре                 | Core Practical                 |    |            |   |  |
| Credits              | 2                              |    |            |   |  |
| No. of Contact Hours | Total Hours                    | 60 | Hours/Week | 4 |  |

| CO-No. | Upon completion of this course, students will be able to:   | PSO addressed | Cognitive Level |
|--------|---|---------------|-----------------|
| CO-1   | Estimate the strength of strong and<br>weak acids, weak bases, FAS and<br>KMnO <sub>4</sub> by conductometric and<br>potentiometric titrations. | PSO-3,4       | Evaluating      |
| CO-2   | Find the dissociation constant of<br>weak acid by conductometric and<br>potentiometric experiments.   | PSO-6,7       | Remembering     |
| CO-3   | Determine the order of saponification of ester by half-life method.   | PSO-8         | Evaluating      |
| CO-4   | Outline the methods of analysis.  | PSO-4         | Understanding   |
| CO-5   | Design the methods for determining<br>the physical constants using<br>electrical instruments.   | PSO-3         | Creating        |

| Course               | Details                  |    |            |   |  |
|----------------------|--------------------------|----|------------|---|--|
| Code                 | 18PCCH21                 |    |            |   |  |
| Title                | INORGANIC CHEMISTRY – II |    |            |   |  |
| Degree               | M. Sc.,                  |    |            |   |  |
| Branch(s)            | Chemistry                |    |            |   |  |
| Year/Semester        | Ι                        | II |            |   |  |
| Туре                 | Core                     |    |            |   |  |
| Credits              | 4                        |    |            |   |  |
| No. of Contact Hours | Total Hours              | 75 | Hours/Week | 5 |  |

| CO-No. | Upon completion of the course, students will able to:                                     | PSO<br>addressed | Cognitive<br>Level           |
|--------|---|------------------|------------------------------|
| CO – 1 | Outline the various reactions in coordination complexes.                                  | PSO-1,2          | Analyzing                    |
| CO - 2 | Interpret the spectra of complexes.   | PSO-6            | Understanding                |
| CO - 3 | Identify the structure of metal carbonyls and clusters.                                   | PSO-2            | Remembering                  |
| CO - 4 | Understand the structure and geometry of interhalogen compounds.                          | PSO-7            | Understanding                |
| CO - 5 | Explain the nuclear fission, fusion<br>reactions and understand the<br>radiation hazards. | PSO-8            | Understanding,<br>Evaluating |

| Course               | Details              |    |            |   |  |
|----------------------|----------------------|----|------------|---|--|
| Code                 | 18PCCH22             |    |            |   |  |
| Title                | ORGANIC CHEMISTRY II |    |            |   |  |
| Degree               | M. Sc.,              |    |            |   |  |
| Branch(s)            | Chemistry            |    |            |   |  |
| Year/Semester        | Ι                    | II |            |   |  |
| Туре                 | Core                 |    |            |   |  |
| Credits              | 4                    |    |            |   |  |
| No. of Contact Hours | Total Hours          | 75 | Hours/Week | 5 |  |

| CO. No. | Upon completion of the course, students will able to:           | PSO<br>addressed | Cognitive<br>Level           |
|---------|---|------------------|------------------------------|
| CO-1    | Explain the basic idea about stereochemistry.                   | PSO-3            | Understanding,<br>Evaluating |
| CO-2    | Find out the mechanism for various name reactions.              | PSO-4            | Remembering                  |
| CO-3    | Outline the reactions of heterocyclic compounds.                | PSO-7            | Understanding                |
| CO-4    | Design the thermal and photochemical reactions.                 | PSO-3            | Creating                     |
| CO-5    | Construct the mechanism for molecular rearrangements reactions. | PSO-3            | Applying,<br>Creating        |

| Course               | Details                 |    |            |   |  |  |
|----------------------|-------------------------|----|------------|---|--|--|
| Code                 | 18PCCH23                |    |            |   |  |  |
| Title                | PHYSICAL CHEMISTRY – II |    |            |   |  |  |
| Degree               | M. Sc.,                 |    |            |   |  |  |
| Branch(s)            | Chemistry               |    |            |   |  |  |
| Year/Semester        | Ι                       | II |            |   |  |  |
| Туре                 | Core                    |    |            |   |  |  |
| Credits              | 4                       |    |            |   |  |  |
| No. of Contact Hours | Total Hours             | 75 | Hours/Week | 5 |  |  |

| CO-No. | Upon completion of this course, students will be able to: | PSO addressed | Cognitive<br>Level           |
|--------|---|---------------|------------------------------|
| CO-1   | Understand the phase rule and colloids.                   | PSO-1,2       | Understanding                |
| CO-2   | Explain the concept of electrochemistry.                  | PSO-3         | Understanding,<br>Evaluating |
| CO-3   | Illustrate the kinetics of electrode reactions.           | PSO-4,5       | Understanding                |
| CO-4   | Construct character table for various point group.        | PSO-8         | Creating                     |
| C0-5   | Apply the group theory to various molecules               | PSO-1,4       | Applying                     |

| Course               | Details           |    |            |   |  |
|----------------------|-------------------|----|------------|---|--|
| Code                 | 18PECH2A          |    |            |   |  |
| Title                | APPLIED CHEMISTRY |    |            |   |  |
| Degree               | M. Sc.,           |    |            |   |  |
| Branch(s)            | Chemistry         |    |            |   |  |
| Year/Semester        | Ι                 | II |            |   |  |
| Туре                 | Elective          |    |            |   |  |
| Credits              | 4                 |    |            |   |  |
| No. of Contact Hours | Total Hours       | 60 | Hours/Week | 4 |  |

| CO-No. | Upon completion of this course, students will be able to:               | PSO addressed | Cognitive<br>Level |
|--------|---|---------------|--------------------|
| CO-1   | Illustrate the types of fuel cells.                                     | PSO-1         | Understanding      |
| CO-2   | Design the techniques of fermentation process.                          | PSO-4         | Creating           |
| CO-3   | Analyse the extraction and properties of oils, fats and waxes.          | PSO-6         | Analysing          |
| CO-4   | Outline the importance and applications of insecticides and pesticides. | PSO-1,2       | Understanding      |
| C0-5   | List out the applications of paints and its functions.                  | PSO-8         | Analysing          |

| Course               | Details         |    |            |   |  |
|----------------------|-----------------|----|------------|---|--|
| Code                 | 18PECH2B        |    |            |   |  |
| Title                | CHEMINFORMATICS |    |            |   |  |
| Degree               | M. Sc.,         |    |            |   |  |
| Branch(s)            | Chemistry       |    |            |   |  |
| Year/Semester        | Ι               | II |            |   |  |
| Туре                 | Elective        |    |            |   |  |
| Credits              | 4               |    |            |   |  |
| No. of Contact Hours | Total Hours     | 60 | Hours/Week | 4 |  |

| CO-No. | Upon completion of the course, students will able to:  | PSO<br>addressed | Cognitive<br>Level           |
|--------|--|------------------|------------------------------|
| CO-1   | Explain the 2d-molecular structure with the aid of computer.                                 | PSO-1,2          | Understanding,<br>Evaluating |
| CO-2   | Understand the characteristics and categories of databases.                                  | PSO-4            | Understanding                |
| CO-3   | Analyze the databases and data source in chemistry.  | PSO-3,6          | Analyzing                    |
| CO-4   | Develop new 3D and chirality descriptors.  | PSO-7            | Creating                     |
| CO-5   | Predict the spectral properties and<br>design combinatorial libraries in<br>cheminformatics. | PSO-8            | Creating                     |

| Course               | Details                         |    |   |            |   |
|----------------------|---------------------------------|----|---|------------|---|
| Code                 | 18PCCH2P1                       |    |   |            |   |
| Title                | ORGANIC CHEMISTRY PRACTICAL – I |    |   |            |   |
| Degree               | M. Sc.,                         |    |   |            |   |
| Branch(s)            | Chemistry                       |    |   |            |   |
| Year/Semester        | Ι                               | Ι  | Ι |            |   |
| Туре                 | Core Practical                  |    |   |            |   |
| Credits              | 2                               |    |   |            |   |
| No. of Contact Hours | Total Hours                     | 60 |   | Hours/Week | 4 |

| CO-No. | Upon completion of the course, students will able to:     | PSO<br>addressed | Cognitive<br>Level |
|--------|---|------------------|--------------------|
| CO-1   | Analyze the organic compounds qualitatively.              | PSO-1,2          | Analyzing          |
| CO-2   | Design the preparation and analysis of organic compounds. | PSO-3,7          | Creating           |
| CO-3   | Predict the functional groups of the organic compounds.   | PSO-3            | Creating           |
| CO-4   | Outline the methods of the analysis.                      | PSO-4            | Understanding      |
| CO-5   | Design the methods for determining the functional groups. | PSO-3            | Creating           |

| Course               | Details                  |    |            |   |   |
|----------------------|--------------------------|----|------------|---|---|
| Code                 | 18PCCH2P2                |    |            |   |   |
| Title                | CHROMATOGRAPHY PRACTICAL |    |            |   |   |
| Degree               | M. Sc.,                  |    |            |   |   |
| Branch(s)            | Chemistry                |    |            |   |   |
| Year/Semester        | Ι                        | II |            |   |   |
| Туре                 | Core Practical           |    |            |   |   |
| Credits              | 2                        |    |            |   |   |
| No. of Contact Hours | Total Hours              | 60 | Hours/Week | C | 4 |

| CO-No. | Upon completion of this course, students will be able to:   | PSO addressed | Cognitive<br>Level           |
|--------|---|---------------|------------------------------|
| CO-1   | Explain the chromatographic separation of organic compounds.  | PSO-4,6       | Understanding,<br>Evaluating |
| CO-2   | Find the $R_f$ values for the separation of cadmium, zinc, red and blue inks.   | PSO-7,8       | Remembering                  |
| CO-3   | Interpret the TLC separation of inorganic compounds.  | PSO-8         | Understanding                |
| CO-4   | Outline the methods of analysis.  | PSO-4         | Understanding                |
| CO-5   | Design the methods for<br>determining the components in the<br>mixture using appropriate<br>chromatographic techniques. | PSO-3         | Creating                     |

| Course               | Details                   |    |     |            |   |
|----------------------|---------------------------|----|-----|------------|---|
| Code                 | 18PCCH31                  |    |     |            |   |
| Title                | INORGANIC CHEMISTRY – III |    |     |            |   |
| Degree               | M. Sc.,                   |    |     |            |   |
| Branch(s)            | Chemistry                 |    |     |            |   |
| Year/Semester        | II                        |    | III |            |   |
| Туре                 | Core                      | ·  |     |            |   |
| Credits              | 4                         |    |     |            |   |
| No. of Contact Hours | Total Hours               | 75 |     | Hours/Week | 5 |

| CO-No. | Upon completion of the course, students will able to:   | PSO<br>addressed | Cognitive<br>Level           |
|--------|---|------------------|------------------------------|
| CO-1   | Outline the preparation and properties of organometallic compounds.   | PSO-1,3          | Understanding                |
| CO-2   | Explain the organic transformations using organo metallic catalyst.   | PSO-2            | Understanding,<br>Evaluating |
| CO-3   | Outline the structure and functional of metalloproteins   | PSO-4,5          | Understanding                |
| CO-4   | Analyze the different structures and reactions of metalloproteins.  | PSO-6,7          | Analyzing                    |
| CO-5   | Explain the synthetic methods,<br>chemical behaviour and<br>characterization of transition metal<br>and hydrogen compounds. | PSO-1,2          | Understanding,<br>Evaluating |

| Course               | Details                 |          |     |            |   |  |
|----------------------|-------------------------|----------|-----|------------|---|--|
| Code                 | 18PCCH32                | 18PCCH32 |     |            |   |  |
| Title                | ORGANIC CHEMISTRY – III |          |     |            |   |  |
| Degree               | M. Sc.,                 |          |     |            |   |  |
| Branch(s)            | Chemistry               |          |     |            |   |  |
| Year/Semester        | II                      |          | III |            |   |  |
| Туре                 | Core                    |          |     |            |   |  |
| Credits              | 4                       |          |     |            |   |  |
| No. of Contact Hours | Total Hours             | 75       |     | Hours/Week | 5 |  |

| CO-No. | Upon completion of the course, students will able to:  | PSO<br>addressed | Cognitive<br>Level           |
|--------|--|------------------|------------------------------|
| CO – 1 | Illustrate the orbital interactions and<br>orbital symmetry correlations of<br>various pericyclic reactions. | PSO-1,2          | Understanding                |
| CO - 2 | Analyze the structure and synthesis route for various alkaloids.   | PSO-3,4          | Analyzing                    |
| CO - 3 | Predict the structures of steroids and terpenoids.   | PSO-5            | Creating                     |
| CO - 4 | Explain the fat and water soluble vitamins.  | PSO-1,2          | Understanding,<br>Evaluating |
| CO - 5 | Design the synthetic route and disconnection approachof various organic molecules.                           | PSO-8            | Creating                     |

| Course               | Details                  |    |     |            |   |
|----------------------|--------------------------|----|-----|------------|---|
| Code                 | 18PCCH33                 |    |     |            |   |
| Title                | PHYSICAL CHEMISTRY – III |    |     |            |   |
| Degree               | M. Sc.,                  |    |     |            |   |
| Branch(s)            | Chemistry                |    |     |            |   |
| Year/Semester        | II                       |    | III |            |   |
| Туре                 | Core                     |    |     |            |   |
| Credits              | 4                        |    |     |            |   |
| No. of Contact Hours | Total Hours              | 75 |     | Hours/Week | 5 |

| CO-<br>No. | Upon completion of this course, students will be able to:       | PSO<br>addressed | Cognitive<br>Level           |
|------------|---|------------------|------------------------------|
| CO-1       | Outline the different photochemical processes.                  | PSO-1            | Understanding                |
| CO-2       | Explain the theory and principle of statistical thermodynamics. | PSO-4            | Understanding,<br>Evaluating |
| CO-3       | Outline the basic concept of statistical thermodynamics         | PSO-1            | Understanding                |
| CO-4       | Apply quantum mechanics concept to hydrogen and helium.         | PSO-1,8          | Applying                     |
| CO-5       | Analyse the various approximation method of quantum chemistry.  | PSO-8            | Analysing                    |

| Course               | Details                          |     |            |   |  |
|----------------------|----------------------------------|-----|------------|---|--|
| Code                 | 18PECH3A                         |     |            |   |  |
| Title                | INSTRUMENTAL METHODS OF ANALYSIS |     |            |   |  |
| Degree               | M. Sc.,                          |     |            |   |  |
| Branch(s)            | Chemistry                        |     |            |   |  |
| Year/Semester        | II                               | III |            |   |  |
| Туре                 | Elective                         | ·   |            |   |  |
| Credits              | 4                                |     |            |   |  |
| No. of Contact Hours | Total Hours                      | 60  | Hours/Week | 4 |  |

| CO-No. | Upon completion of this course, students will be able to:  | PSO<br>addressed | Cognitive<br>Level           |
|--------|--|------------------|------------------------------|
| CO-1   | Explain the principle and application of thermo-analytical methods.  | PSO-1            | Understanding,<br>Evaluating |
| CO-2   | Analyze the various electro-<br>analytical methods   | PSO-2            | Analyzing                    |
| CO-3   | Explain the spectrophotometric analysis and its applications.  | PSO-4            | Understanding,<br>Evaluating |
| CO-4   | Illustratetheprinciple,instrumentation and applications ofFlurometry,Flamephotometry,Nephelometry and Atomic absorptionSpectroscopy. | PSO-1            | Understanding,<br>Evaluating |
| CO-5   | Outline the principle and<br>instrumentation of NMR, PES and<br>Medical Imaging techniques.  | PSO-1,8          | Understanding                |

| Course               | Details          |          |     |            |   |  |
|----------------------|------------------|----------|-----|------------|---|--|
| Code                 | 18PECH3B         | 18PECH3B |     |            |   |  |
| Title                | ENZYME CHEMISTRY |          |     |            |   |  |
| Degree               | M. Sc.,          |          |     |            |   |  |
| Branch(s)            | Chemistry        |          |     |            |   |  |
| Year/Semester        | II               |          | III |            |   |  |
| Туре                 | Elective         |          |     |            |   |  |
| Credits              | 4                |          |     |            |   |  |
| No. of Contact Hours | Total Hours      | 60       |     | Hours/Week | 4 |  |

| CO-No. | Upon completion of this course, students will be able to: | PSO addressed | Cognitive<br>Level |
|--------|---|---------------|--------------------|
|        | Explain the classification,                               |               |                    |
| CO-1   | nomenclature and purification of                          | PSO-6         | Understanding,     |
|        | enzymes.  |               | Evaluating         |
|        | Design the kinetics and mechanism                         |               |                    |
| CO-2   | of enzyme catalyzed reactions.                            | PSO-1,2       | Creating           |
|        | Analyze the mechanism of enzyme                           |               |                    |
| CO-3   | catalyzed reactions.                                      | PSO-4         | Analyzing          |
|        | Evaluate the structure and                                |               |                    |
| CO-4   | function of multiple enzyme                               | PSO-7         | Evaluating         |
|        | complexes.  |               | Dvardating         |
|        | Outline the industrial application of                     |               |                    |
| CO-5   | extremozymes.   | PSO-8         | Understanding      |

| Course               | Details                          |    |            |   |  |
|----------------------|----------------------------------|----|------------|---|--|
| Code                 | 18PCCH3P1                        |    |            |   |  |
| Title                | ORGANIC CHEMISTRY PRACTICAL – II |    |            |   |  |
| Degree               | M. Sc.,                          |    |            |   |  |
| Branch(s)            | Chemistry                        |    |            |   |  |
| Year/Semester        | II                               | Ι  | II         |   |  |
| Туре                 | Core Practical                   |    |            |   |  |
| Credits              | 2                                |    |            |   |  |
| No. of Contact Hours | Total Hours                      | 60 | Hours/Week | 4 |  |

| CO-No. | Upon completion of the course, students will able to:     | PSO<br>addressed | BT                      |
|--------|---|------------------|-------------------------|
| CO-1   | Outline the preparation of organic compounds.             | PSO-2,4          | Understanding           |
| CO-2   | Analyze the different stages of preparation of compounds. | PSO-5,7          | Analyzing               |
| CO-3   | Design new methods of synthesis.                          | PSO-3            | Creating                |
| CO-4   | Outline the methods of analysis.                          | PSO-4            | Understanding           |
| CO-5   | Estimate the organic compounds using different methods.   | PSO-3            | Evaluating,<br>Creating |

| Course               | Details                         |    |     |            |   |
|----------------------|---------------------------------|----|-----|------------|---|
| Code                 | 18PCCH3P2                       |    |     |            |   |
| Title                | PHYSICAL CHEMISTRY PRACTICAL II |    |     |            |   |
| Degree               | M. Sc.,                         |    |     |            |   |
| Branch(s)            | Chemistry                       |    |     |            |   |
| Year/Semester        | II                              |    | III |            |   |
| Туре                 | Core Practical                  |    |     |            |   |
| Credits              | 2                               |    |     |            |   |
| No. of Contact Hours | Total Hours                     | 60 |     | Hours/Week | 4 |

| CO-No. | Upon completion of this course, students will be able to:  | PSO addressed | Cognitive<br>Level           |
|--------|--|---------------|------------------------------|
| CO-1   | Analyze Ostwald's dilution law and<br>Primary salt effect.   | PSO-1,2       | Analyzing                    |
| CO-2   | Explain the kinetics of<br>Persulphate-iodide reaction.  | PSO-4         | Understanding,<br>Evaluating |
| CO-3   | Determine the heat of solution of<br>naphthalene – toluene, oxalic acid<br>– water and ammonium oxalate –<br>water systems.                | PSO-8         | Evaluating                   |
| CO-4   | Analyze the adsorption process<br>and determine the unknown<br>concentration.  | PSO-7         | Analyzing                    |
| CO-5   | Estimate the partial molar volume<br>of solute and analyze the<br>stoichiometry, stability constant of<br>inorganic and organic complexes. | PSO-8         | Evaluating,<br>Creating      |

| Course               | Details      |    |    |            |   |
|----------------------|--------------|----|----|------------|---|
| Code                 | 18PCCH41     |    |    |            |   |
| Title                | SPECTROSCOPY |    |    |            |   |
| Degree               | M. Sc.,      |    |    |            |   |
| Branch(s)            | Chemistry    |    |    |            |   |
| Year/Semester        | II           |    | IV |            |   |
| Туре                 | Core         | 1  |    |            |   |
| Credits              | 4            |    |    |            |   |
| No. of Contact Hours | Total Hours  | 75 |    | Hours/Week | 5 |

| CO-No. | Upon completion of the course, students will able to: | PSO<br>addressed | Cognitive<br>Level |
|--------|---|------------------|--------------------|
| CO - 1 | Analyze the spectrum of UV and                        |                  | A 1 .              |
|        | IR spectroscopy.                                      | PSO-1,3          | Analyzing          |
| CO – 2 | Interpret the structure of compounds                  |                  | Understanding,     |
|        | using NMR spectroscopy.                               | P50-4            | Evaluating         |
| CO 2   | Analyze the applications of Mass                      |                  |                    |
| 0-5    | spectrometry.   | PSO-6            | Analyzing          |
|        | Outline the properties and                            |                  |                    |
| CO – 4 | principle of ORD and CD.                              | PSO-7            | Analyzing          |
|        | Assess the structure of inorganic                     |                  |                    |
| CO – 5 | complexes using NMR, EPR and                          | PSO-8            | Evaluating         |
|        | Mössbauer spectroscopy.                               |                  |                    |

| Course               | Details                      |    |  |            |   |  |
|----------------------|------------------------------|----|--|------------|---|--|
| Code                 | 18PCCH42                     |    |  |            |   |  |
| Title                | ADVANCED TOPICS IN CHEMISTRY |    |  |            |   |  |
| Degree               | M. Sc.,                      |    |  |            |   |  |
| Branch(s)            | Chemistry                    |    |  |            |   |  |
| Year/Semester        | II IV                        |    |  |            |   |  |
| Туре                 | Core                         |    |  |            |   |  |
| Credits              | 4                            |    |  |            |   |  |
| No. of Contact Hours | Total Hours                  | 75 |  | Hours/Week | 5 |  |

| CO-No. | Upon completion of this course, students will be able to:       | PSO addressed | Cognitive<br>Level           |
|--------|---|---------------|------------------------------|
| CO-1   | Explain the basic concept of green chemistry.                   | PSO-1,2       | Understanding,<br>Evaluating |
| CO-2   | Outline the basic concept of nanochemistry.                     | PSO-2         | Understanding                |
| CO-3   | Analyze the concept of supramolecular chemistry.                | PSO-4         | Analyzing                    |
| CO-4   | Apply the concept of<br>thermodynamics in biological<br>systems | PSO-7         | Applying                     |
| CO-5   | Apply the database for drug designing.                          | PSO-8         | Applying                     |

| Course               | Details     |     |            |   |
|----------------------|-------------|-----|------------|---|
| Code                 | 18PCCH43    |     |            |   |
| Title                | PROJECT     |     |            |   |
| Degree               | M. Sc.,     |     |            |   |
| Branch(s)            | Chemistry   |     |            |   |
| Year/Semester        | II          | IV  |            |   |
| Туре                 | Core        | ·   |            |   |
| Credits              | 8           |     |            |   |
| No. of Contact Hours | Total Hours | 120 | Hours/Week | 8 |

| CO-No. | Upon completion of this course, students will be able to:                          | PSO addressed | Cognitive<br>Level |
|--------|--|---------------|--------------------|
| CO-1   | Design new hypothesis, carry out<br>experimental works and analyze<br>the results. | PSO-3         | Creating           |
| CO-2   | Analyze the scientific problems and find novel solutions.                          | PSO-4         | Analyzing          |
| CO-3   | Apply critical thinking and<br>analytical reasoning to scientific<br>problems.     | PSO-8         | Applying           |
| CO-4   | Outline the results in oral, written and electronic formats.                       | PSO-8         | Understanding      |
| CO-5   | Experiment with new ideas and new areas of research.                               | PSO-8         | Applying           |

| Course               | Details             |    |            |   |  |  |
|----------------------|---------------------|----|------------|---|--|--|
| Code                 | 18PECH4A            |    |            |   |  |  |
| Title                | MEDICINAL CHEMISTRY |    |            |   |  |  |
| Degree               | M. Sc.,             |    |            |   |  |  |
| Branch(s)            | Chemistry           |    |            |   |  |  |
| Year/Semester        | II                  | IV |            |   |  |  |
| Туре                 | Elective            |    |            |   |  |  |
| Credits              | 4                   |    |            |   |  |  |
| No. of Contact Hours | Total Hours         | 60 | Hours/Week | 4 |  |  |

| CO-No. | Upon completion of the course, students will able to:                       | PSO<br>addressed | Cognitive<br>Level           |
|--------|---|------------------|------------------------------|
| CO-1   | Explain the nomenclature and terminology in drugs.                          | PSO-1            | Understanding,<br>Evaluating |
| CO-2   | Identify the mode of action of sedatives and hypnotics.                     | PSO-2,3          | Applying                     |
| CO-3   | Explain the mechanism of autonomic drugs.                                   | PSO-5            | Understanding,<br>Evaluating |
| CO-4   | Illustrate the structure and functional of diuretics                        | PSO-7            | Understanding                |
| CO-5   | Elaborate the mechanism of<br>antihistamine and anti-parkinsonism<br>drugs. | PSO-8            | Creating                     |

| Course               | Details              |    |    |            |   |  |
|----------------------|----------------------|----|----|------------|---|--|
| Code                 | 18PECH4B             |    |    |            |   |  |
| Title                | RATIONAL DRUG DESIGN |    |    |            |   |  |
| Degree               | M. Sc.,              |    |    |            |   |  |
| Branch(s)            | Chemistry            |    |    |            |   |  |
| Year/Semester        | II                   |    | IV |            |   |  |
| Туре                 | Elective             |    |    |            |   |  |
| Credits              | 4                    |    |    |            |   |  |
| No. of Contact Hours | Total Hours          | 60 |    | Hours/Week | 4 |  |

| CO-No. | Upon completion of the course, students will able to:                                     | PSO<br>addressed | ВТ            |
|--------|---|------------------|---------------|
| CO – 1 | Identify the different routes of drug administration.                                     | PSO-3            | Applying      |
| CO – 2 | Illustrate the action of drugs.   | PSO-5            | Understanding |
| CO – 3 | Outline the chemical structure and<br>pharmacologic activity in a<br>quantitative manner. | PSO-8            | Understanding |
| CO – 4 | Designthe binding affinity between molecules in drug design.                              | PSO-8            | Creating      |
| CO – 5 | Interpret the interaction of drugs within living organisms.                               | PSO-7            | Evaluating    |

| Course               | Details                          |    |              |  |  |  |
|----------------------|----------------------------------|----|--------------|--|--|--|
| Code                 | 18PCCH4P1                        |    |              |  |  |  |
| Title                | INORGANIC CHEMISTRY PRACTICAL II |    |              |  |  |  |
| Degree               | M. Sc.,                          |    |              |  |  |  |
| Branch(s)            | Chemistry                        |    |              |  |  |  |
| Year/Semester        | II                               | ]  | IV           |  |  |  |
| Туре                 | Core Practical                   |    |              |  |  |  |
| Credits              | 2                                |    |              |  |  |  |
| No. of Contact Hours | Total Hours                      | 60 | Hours/Week 4 |  |  |  |

| CO-No. | Upon completion of the course, students will able to:                       | PSO<br>addressed | ВТ                           |
|--------|---|------------------|------------------------------|
| CO-1   | Outline the preparation of inorganic complexes.                             | PSO-3            | Understanding                |
| CO-2   | Explain the volumetric analysis of inorganic elements.                      | PSO-5            | Understanding,<br>Evaluating |
| CO-3   | Design the methods of estimation of elements.                               | PSO-8            | Creating                     |
| CO-4   | Outline the methods of qualitative analysis.                                | PSO-4            | Understanding                |
| CO-5   | Design the effective gravimetric estimation methods for inorganic elements. | PSO-3            | Creating                     |

| Course               | Details                   |    |    |            |   |  |
|----------------------|---------------------------|----|----|------------|---|--|
| Code                 | 18PCCH4P2                 |    |    |            |   |  |
| Title                | GREEN CHEMISTRY PRACTICAL |    |    |            |   |  |
| Degree               | M. Sc.,                   |    |    |            |   |  |
| Branch(s)            | Chemistry                 |    |    |            |   |  |
| Year/Semester        | II                        |    | IV |            |   |  |
| Туре                 | Core Practical            |    |    |            |   |  |
| Credits              | 2                         |    |    |            |   |  |
| No. of Contact Hours | Total Hours               | 60 |    | Hours/Week | 4 |  |

| CO-No. | Upon completion of this course, students will be able to:                               | PSO<br>addressed | вт                      |
|--------|---|------------------|-------------------------|
| CO-1   | Outline the preparation of organic<br>and inorganic compounds using<br>green chemistry. | PSO-7,8          | Understanding           |
| CO-2   | Make use of green techniques to identify radicals.                                      | PSO-8            | Applying                |
| CO-3   | Test N, S, Cl, Br and I using green chemistry.  | PSO-6            | Creating,<br>Evaluating |
| CO-4   | Design new methods of preparation<br>of organic compounds using green<br>chemistry.     | PSO-3            | Creating                |
| CO-5   | Categorize the chemicals as hazardous and safer to use.                                 | PSO-4            | Analyzing               |

| Course               | Details              |    |    |            |   |
|----------------------|----------------------|----|----|------------|---|
| Code                 | 18PICH21             |    |    |            |   |
| Title                | INDUSTRIAL CHEMISTRY |    |    |            |   |
| Degree               | M. Sc.,              |    |    |            |   |
| Branch(s)            | Chemistry            |    |    |            |   |
| Year/Semester        | Ι                    |    | II |            |   |
| Туре                 | IDC                  |    |    |            |   |
| Credits              | 3                    |    |    |            |   |
| No. of Contact Hours | Total Hours          | 45 |    | Hours/Week | 3 |

| CO-No. | Upon completion of this course, students will be able to:                                   | PSO addressed | ВТ                           |  |
|--------|---|---------------|------------------------------|--|
| CO-1   | Explain the extraction, properties and uses of different metals.                            | PSO-1,3       | Understanding,<br>Evaluating |  |
| CO-2   | Outline the manufacturing process of cement.  | PSO-5         | Understanding                |  |
| CO-3   | Outline the composition<br>preparation and uses of matchbox,<br>explosives and propellants. | PSO-4         | Understanding                |  |
| CO-4   | Analyse the preparation of day to day articles.   | PSO-4         | Analysing                    |  |
| CO-5   | Outline the synthesis and properties of petrochemicals.                                     | PSO-2         | Understanding                |  |

| Course               | Details                         |    |     |            |   |
|----------------------|---------------------------------|----|-----|------------|---|
| Code                 | 18PICH31                        |    |     |            |   |
| Title                | INTRODUCTION TO CHEMINFORMATICS |    |     |            |   |
| Degree               | M. Sc.,                         |    |     |            |   |
| Branch(s)            | Chemistry                       |    |     |            |   |
| Year/Semester        | II                              |    | III |            |   |
| Туре                 | IDC                             |    |     |            |   |
| Credits              | 3                               |    |     |            |   |
| No. of Contact Hours | Total Hours                     | 45 |     | Hours/Week | 3 |

| CO-No. | Upon completion of the course, students will able to:   | PSO<br>addressed | ВТ                           |
|--------|---|------------------|------------------------------|
| CO-1   | Explain the 2D, 3D molecular<br>structures and basics of<br>cheminformatics.                                    | PSO-1,2          | Understanding,<br>Evaluating |
| CO-2   | Classify the different database and data sources in chemistry.  | PSO-4            | Analyzing                    |
| CO-3   | Outline the applications of chemical information searches.  | PSO-5            | Understanding                |
| CO-4   | Assess the various types of molecular descriptor.   | PSO-8            | Evaluating                   |
| CO-5   | Apply novel methods in drug<br>designing and analyzing the<br>pharmacokinetics action of drug<br>on human body. | PSO-8            | Applying                     |